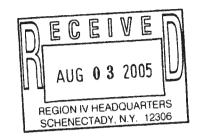


136 Coonbrook Road • P.O. Box 69 Petersburgh, New York 12138 Telephone: 518.658.3202

Fax: 518.658.3204

August 3, 2005

Mr. Frederick Sievers
Regional Water Engineer
New York State Department of Environmental Conservation
Region 4
1150 North Westcott Road
Schenectady, NY 12306-2014



Re: Taconic, Petersburgh, NY

Dear Mr. Sievers:

We have recently undertaken voluntary efforts to evaluate the presence and level of perfluorooctanic acid (PFOA) at or near the Taconic facility. Attached please find copies of the laboratory analytical reports related to the presence of this compound in groundwater at or near the facility.

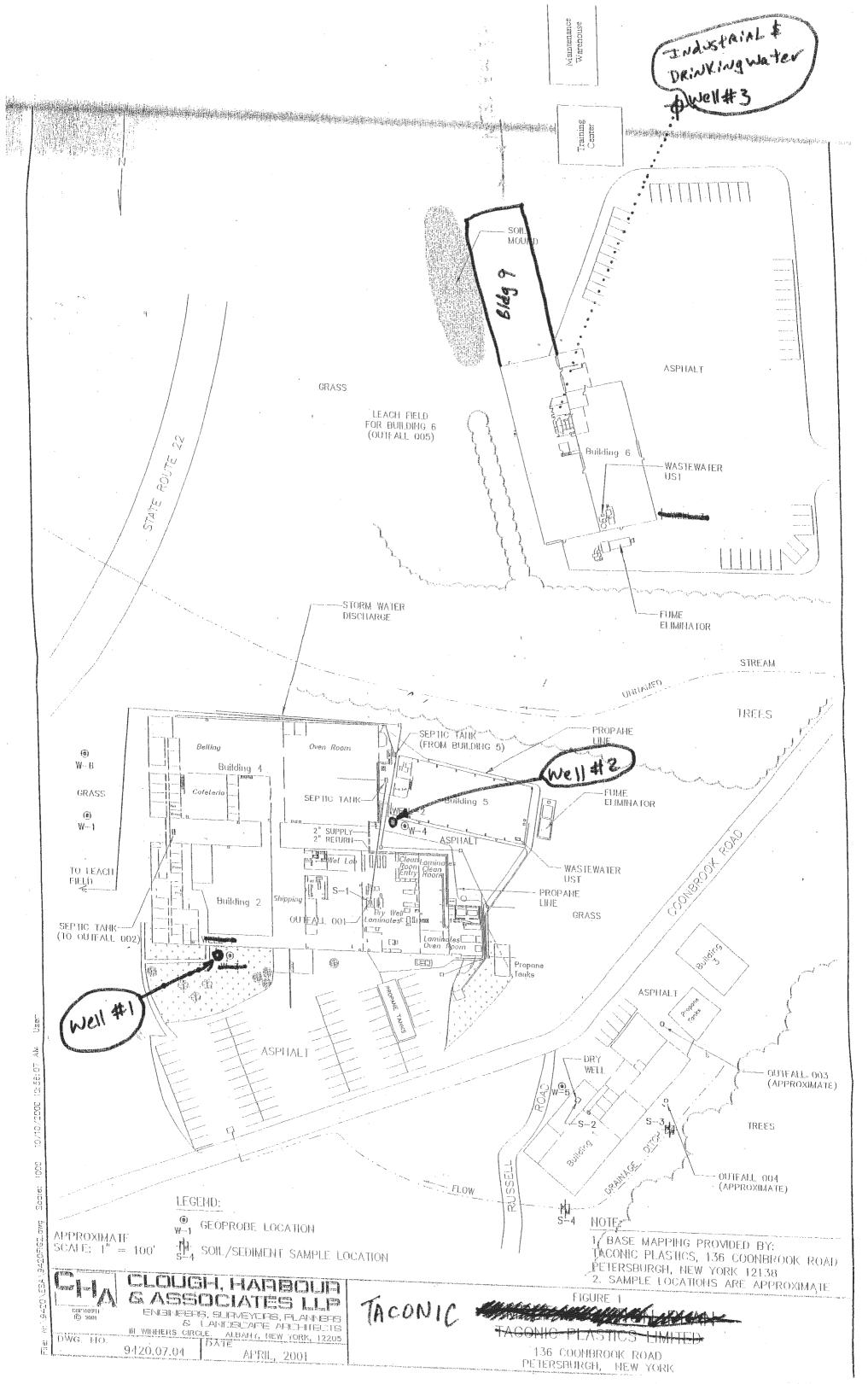
After you have had a chance to look through the enclosed, would you please contact me at your earliest convenience so that we can arrange a meeting to discuss this information? I can be reached at (518) 658-3202 ext. 288.

Sincerely,

Andrew Kawczak

Environmental Manager

DECinfoLetter







Analytical Report

TACONIC Analysis of PFOA in Water Samples Exygen Report No. L0003924

Testing Laboratory

Exygen Research 3058 Research Drive State College, PA 16801

Requester

Tim Kosto TACONIC 136 Coonbrook Road Petersburgh, New York, 12138 Phone: 518-658-3202

> See Mge 12 ±



1 Introduction

Results are reported for the analysis of perfluorooctanoic acid (PFOA) in wastewater samples received at Exygen from Tim Kosto at TACONIC. The Exygen project number assigned to the samples is L0003924. Table I lists the target analytes quantitated for the samples.

Table I. Target Analytes for Quantitation

<u>Parameter</u>	<u>Acronym</u>	<u>Formula</u>
Perfluorooctanoic acid	PFOA	C ₇ F ₁₅ COOH

2 Sample Receipt

Four samples were received at Exygen in 500 mL clear plastic bottles. A copy of all sample log-in information is presented in Attachment A.

The samples were received on 11/23/04. The samples were shipped on ice via UPS. The samples were stored refrigerated from time of receipt until analysis.

3 Methods - Analytical and Preparatory

3.1 Sample Preparation

Solid phase extraction (SPE) was used to prepare the samples for LC/MS/MS analysis. A forty milliliter portion of each sample was transferred to a C_{18} SPE cartridge. The cartridge was eluted with 5 mL of 100% methanol. This treatment resulted in an eight-fold concentration of the samples prior to analysis. A portion of the extract was transferred to autosampler vials and analyzed using electrospray LC/MS/MS.

3.2 Sample Analysis by LC/MS/MS

In High Pressure Liquid Chromatography (HPLC), an aliquot of extract is injected and passed through a liquid-phase chromatographic column. Based on the affinity of the analyte for the stationary phase in the column relative to the liquid mobile phase, the analyte is retained for a characteristic amount of time. Following HPLC separation, mass spectrometry provides a rapid and accurate means for analyzing a wide range of organic compounds. Molecules are ionized, fragmented, and detected. The ions characteristic of the compounds are observed and quantitated against extracted standards.

An HP1100 system interfaced to a Micromass Quattro system was used to analyze the sample extracts for quantitation. A gradient elution through a Jones Chromatography Genesis C-8 50 x 2.1 mm x 4µm column was used for separation.

The following gradient was performed:

Mobile Phase (A): Mobile Phase (B):	2mM Am Methanol	monium Acetate in Water
<u>Time</u>	<u>%A</u>	%B
0.0	60	40
0.4	60	40
1.0	10	90
7.0	1Q	90
7.5	0	100
9.0	0	100
9.5	60	40
13.5	60	40
14.0	60	40

The following parameters were used for operation of the mass spectrometer:

Setting
Electrospray
Negative
413->369 (PFOA)
350°C
7.0 L/min

4 Analysis

4.1 Calibration

A 7-point calibration curve was analyzed throughout the analytical sequence for PFOA. The calibration points were prepared at 0, 25, 50, 100, 250, 500, and 1000 ng/L (ppt) for LC/MS/MS analysis. Calibration standards are prepared using the same SPE procedure used for samples. The instrument response versus the concentration was plotted for each point. Using linear regression with 1/x weighting, the slope, y-intercept and coefficient of determination (r^2) were determined. A calibration curve is acceptable if $r^2 \ge 0.985$.

For the results reported here, calibration criteria were met. The calibration curves are included in the raw data in Attachment C.

4.2 Surrogates

Surrogates were not used in this study.

4.3 Laboratory Control Spikes

Laboratory control spikes in the analytical set were prepared by adding a known concentration of the analyte to laboratory water. Laboratory control spikes are used to assess method accuracy. The laboratory control spikes must show recoveries between 70-130% or the data is rejected. For the results reported here, the laboratory control spikes were within the acceptable range.

4.4 Matrix Spikes

A matrix spike was prepared for each sample in the analytical set by adding a known concentration of the target analyte to a separate sample. Matrix spikes are used to assess method accuracy in the matrix. The matrix spikes should show recoveries between 70-130%. For the results reported here the matrix spike for 04-10-03A,B was within the acceptable range. For the other three samples, 04-10-01A,B, 04-10-02A,B and 04-10-04, the amount of PFOA found in the sample greatly exceeded the spiking level and an accurate recovery could not be calculated.

4.5 Sample Related Comments

Each sample was extracted in duplicate and analyzed. Duplicate sample results are reported along with the sample results in Attachment B.

5 Data Summary

Please see Attachment B for a detailed listing of the analytical results. Results are reported in parts per billion (ng/mL) for the analyte, PFOA, on an as-received basis.

6 Data/Sample Retention

Samples are disposed of one month after the report is issued unless otherwise specified. All electronic data is archived on retrievable media and hard copy reports are stored in data folders maintained by Exygen. Hardcopy data is stored for a minimum of five years. The client will be notified 30 days prior to the disposal of hardcopy data.

7 Attachments

- 7.1 Attachment A: Chain of Custody
- 7.2 Attachment B: Analytical Results
- 7.3 Attachment C: Raw Analytical Data

Signatures	
Saculisha	12/21/04
Karen Risha, Principal Investigator	Date
Ool M Fland	12/21/09
John M. Flaherty, Vice President	Date

Additioned Suples For PECA testing of TExTERN: 04-10-03 A Bldg 6 Dernikuz WARRE 04-10-04 PFOA CONTROL WATTRE (DISHILL) 18 , Distilled Water Wale 415.2 Wahr FC 2.11.22 9 1:100 (Jefer





Summary of PFOA in Water Samples

Sample ID	Analyte Found ppb (ng/mL) PFOA
04-10-01A,B	117 Bld 5 44 (44.4 42) 116) Bld 5 44 (44.4 42)
04-10-01A,B*	116 Jule 19 Charles
04-10-02A,B	152 7 3 3 3 4 5 4 6 7 7 7 7
04-10-02A,B*	2.30 7 31 dg # 6 22 11 (A-RA, 2001) # 3)
04-10-03A,B	2.30 7 20 10 16 10 10 10 10 10 10 10 10 10 10 10 10 10
04-10 - 03A,B*	2.07
04-10-04	13700
04-10-04*	14100

^{*}Laboratory Duplicate



January 2005

Analytical Report

TACONIC Analysis of PFOA in Water Samples Exygen Report No. L0004258

Testing Laboratory

Exygen Research 3058 Research Drive State College, PA 16801

Requester

Tim Kosto TACONIC 136 Coonbrook Road Petersburgh, New York, 12138 Phone: 518-658-3202

See page 11 =

3058 Research Drive State College, PA 16801, USA T: 800.281.3219 F: 814.272.1019 exygen.com

1 Introduction

Results are reported for the analysis of perfluorooctanoic acid (PFOA) in water samples received at Exygen from Tim Kosto at TACONIC. The Exygen project number assigned to the samples is L0004258. Table I lists the target analytes quantitated for the samples.

Table I. Target Analytes for Quantitation

<u>Parameter</u>	<u>Acronym</u>	<u>Formula</u>
Perfluorooctanoic acid	PFOA	C ₇ F ₁₅ COOH

2 Sample Receipt

Five samples were received at Exygen in 500 mL clear plastic bottles. A copy of all sample log-in information is presented in Attachment A.

The samples were received on 01/14/05. The samples were shipped on ice via UPS. The samples were stored refrigerated from time of receipt until analysis.

3 Methods - Analytical and Preparatory

3.1 Sample Preparation

Solid phase extraction (SPE) was used to prepare the samples for LC/MS/MS analysis. A forty milliliter portion of each sample was transferred to a C_{18} SPE cartridge. The cartridge was eluted with 5 mL of 100% methanol. This treatment resulted in an eight-fold concentration of the samples prior to analysis. A portion of the extract was transferred to autosampler vials and analyzed using electrospray LC/MS/MS.

3.2 Sample Analysis by LC/MS/MS

In High Pressure Liquid Chromatography (HPLC), an aliquot of extract is injected and passed through a liquid-phase chromatographic column. Based on the affinity of the analyte for the stationary phase in the column relative to the liquid mobile phase, the analyte is retained for a characteristic amount of time. Following HPLC separation, mass spectrometry provides a rapid and accurate means for analyzing a wide range of organic compounds. Molecules are ionized, fragmented, and detected. The ions characteristic of the compounds are observed and quantitated against extracted standards.

An HP1100 system interfaced to a Micromass Quattro system was used to analyze the sample extracts for quantitation. A gradient elution through a Jones Chromatography Genesis C-8 $50 \times 2.1 \text{ mm} \times 4 \mu \text{m}$ column was used for separation:

The following gradient was performed:

Mobile Phase (A): Mobile Phase (B):	2mM Ammonium Methanol	n Acetate in Water
Time 0.0 0.4 1.0 7.0 7.5 9.0 9.5 13.5 14.0	%A 60 60 10 10 0 0 60 60	%B 40 40 90 90 100 100 40 40

The following parameters were used for operation of the mass spectrometer:

_	The second second contract of the second cont
The second secon	Setting
Parameter	Electrospray
Ionization Mode	Negative
Polarity	413->369 (PFOA)
Transitions Monitored	350°C
Gas Temperature	000 0
Daving Cae (N2)	7.0 L/min
DIVING GAS INC.	Application measures

4 Analysis

4.1 Calibration

A 7-point calibration curve was analyzed throughout the analytical sequence for PFOA. The calibration points were prepared at 0, 25, 50, 100, 250, 500, and 1000 ng/L (ppt) for LC/MS/MS analysis. Calibration standards are prepared using the same SPE procedure used for samples. The instrument response versus the concentration was plotted for each point. Using linear regression with 1/x weighting, the slope, y-intercept and coefficient of determination (r^2) were determined. A calibration curve is acceptable if $r^2 \ge 0.985$.

For the results reported here, calibration criteria were met. The calibration curves are included in the raw data in Attachment C.

4.2 Surrogates

Surrogates were not used in this study.

4.3 Laboratory Control Spikes

Laboratory control spikes in the analytical set were prepared by adding a known concentration of the analyte to laboratory water. Laboratory control spikes are used to assess method accuracy. The laboratory control spikes must show recoveries between 70-130% or the data is rejected. For the results reported here, the laboratory control spikes were within the acceptable range.

4.4 Matrix Spikes

A matrix spike was prepared for each sample in the analytical set by adding a known concentration of the target analyte to a separate sample. Matrix spikes are used to assess method accuracy in the matrix. The matrix spikes should show recoveries between 70-130%. For the results reported here the matrix spike for 04-10-03A,B was within the acceptable range. For the other three samples, 04-10-01A,B, 04-10-02A,B and 04-10-04, the amount of PFOA found in the sample greatly exceeded the spiking level and an accurate recovery could not be calculated.

4.5 Sample Related Comments

Each sample was extracted in duplicate and analyzed. Duplicate sample results are reported along with the sample results in Attachment B.

5 Data Summary

Please see Attachment B for a detailed listing of the analytical results. Results are reported in parts per billion (ng/mL) for the analyte, PFOA.

6 Data/Sample Retention

Samples are disposed of one month after the report is issued unless otherwise specified. All electronic data is archived on retrievable media and hard copy reports are stored in data folders maintained by Exygen. Hardcopy data is stored for a minimum of five years. The client will be notified 30 days prior to the disposal of hardcopy data.

7 Attachments

- 7.1 Attachment A: Chain of Custody
- 7.2 Attachment B: Analytical Results
- 7.3 Attachment C: Raw Analytical Data

Muly Ma	01/24/05
Karen Risha, Principal Investigator	Date
alm Flakast	1/34/05
John M. Flaherty, Vice President	Date

8 Signatures

1/13/05		
- American de la companya del la companya de la com	4	
PFOA	ANALYSIS	
	· · · · · · · · · · · · · · · · · · ·	

SAMPLE ID	LOCATION	
04-13-01	RPI Described Weder	
04-13-02	161 SHEFAUT ROAD	ملا
04-13-03	147 Coompaoon Rp (Lon	4.3
04-13-04	6 RUSSALL ROAD	6.2)
04-13-05	BUILDING 1	(< 6

4/13/05

FLUOROSILICONE EURLUMTIONS FUR GASKO

DOW CORNING LS 5-8761 FLUOROSIZICONA

20,09 L3 5-876/ Lot 000 2224489 20,189 2-BUTYDNONN SIAMS LOTTE 0095 ZTC

20.71 q 65 5-876/ Lot 000 2224489 20.85 g 4-mily 2-penter Bthe 00955HC

BOTH Sompras Principano At ~ 10:30 1.M.

AFTER 2 hours, both solutions had partially solvated.

sticking with a tough depressor resulted in Schinding and a thicker ball of solvated shurrosilizer or the solvated





Summary of PFOA in Water Samples

Sample ID	Analyte Found ppb (ng/mL) PFOA
The state of the s	
04-13-01	ND 7
04-13-01*	ND)
04-13-02	ND 3 161 5 m 3 - 1 trad ND 3 161 5 m 3 - 21 trad ND 3 161 5 m 3 - 21 trad ND 3 2 dack /4/455 dack
04-13-02*	ND) Scalack /NASSAL
04-13-03	4.20 } 147 coord 8 rook &
04-13-03*	
04-13-04	2.28 } 6 & 2352\ P. J
04-13-04*	
04-13-05	0.562 } \$1d \$ \$1 (vid pand)
04-13-05*	0.516 3

*Laboratory Duplicate

ND = Not Detected. Result is less than 0.0500 ng/mL.

